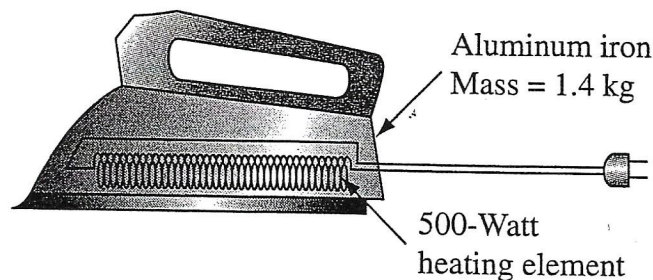


## ASSIGNMENT #5

### Lumped System Analysis Problems

- 5-1 A 0.6-cm-diameter mild steel rod at  $38^\circ\text{C}$  is suddenly immersed in a liquid at  $93^\circ\text{C}$  with  $\bar{h}_c = 110 \text{ W/m}^2 \text{ K}$ . Determine the time required for the rod to warm to  $88^\circ\text{C}$ .
- 5-2 A thin-wall cylindrical vessel (1 m in diameter) is filled to a depth of 1.2 m with water at an initial temperature of  $15^\circ\text{C}$ . The water is well stirred by a mechanical agitator. Estimate the time required to heat the water to  $50^\circ\text{C}$  if the tank is suddenly immersed in oil at  $105^\circ\text{C}$ . The overall heat transfer coefficient between the oil and the water is  $284 \text{ W/m}^2 \text{ K}$ , and the effective heat transfer surface is  $4.2 \text{ m}^2$ .
- 5-3 A large, 2.54-cm.-thick copper plate is placed between two air streams. The heat transfer coefficient on one side is  $28 \text{ W/m}^2 \text{ K}$  and on the other side is  $57 \text{ W/m}^2 \text{ K}$ . If the temperature of both streams is suddenly changed from  $38^\circ\text{C}$  to  $93^\circ\text{C}$ , determine how long it will take for the copper plate to reach a temperature of  $82^\circ\text{C}$ .
- 5-4 A 1.4-kg aluminum household iron has a 500-W heating element. The surface area is  $0.046 \text{ m}^2$ . The ambient temperature is  $21^\circ\text{C}$ , and the surface heat transfer coefficient is  $11 \text{ W/m}^2 \text{ K}$ . How long after the iron is plugged in will its temperature reach  $104^\circ\text{C}$ ?



5-5

Ball bearings are to be hardened by quenching them in a water bath at a temperature of  $37^\circ\text{C}$ . You are asked to devise a continuous process in which the balls could roll from a soaking oven at a uniform temperature of  $870^\circ\text{C}$  into the water, where they are carried away by a rubber conveyor belt. The rubber conveyor belt, however, would not be satisfactory if the surface temperature of the balls leaving the water is above  $90^\circ\text{C}$ . If the surface coefficient of heat transfer between the balls and the water can be assumed to be equal to  $590 \text{ W/m}^2 \text{ K}$ , (a) find an approximate relation giving the minimum allowable cooling time in the water as a function of the ball radius for balls up to  $1.0 \text{ cm}$  in diameter, (b) calculate the cooling time, in seconds, required for a ball having a  $2.5\text{-cm}$  diameter, and (c) calculate the total amount of heat in watts that would have to be removed from the water bath in order to maintain a uniform temperature if  $100,000$  balls of  $2.5\text{-cm}$  diameter are to be quenched per hour.

